

HORMOTHAMNIONE, A NOVEL CYTOTOXIC STYRYLCHROMONE FROM THE MARINE CYANOPHYTE
HORMOTHAMNION ENTEROMORPHOIDES GRUNOW

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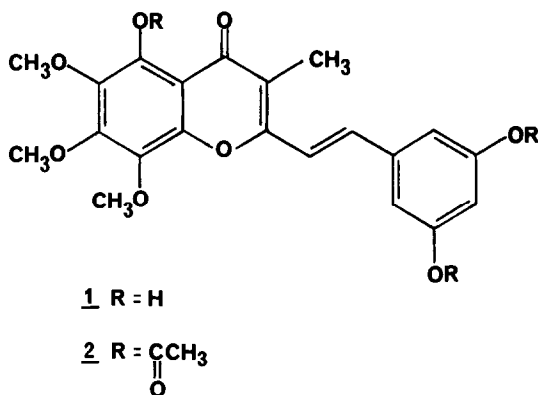
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Abstract: The first naturally occurring styrylchromone, hormothamnione, was isolated from the marine cyanophyte Hormothamnion enteromorphoides and its structure determined by an x-ray experiment on its triacetate derivative. Hormothamnione is an exceptionally potent cytotoxin to cancer cells in vitro and appears to be a selective inhibitor of RNA synthesis.

Blue green algae have recently received considerable attention by academic researchers,^{1,2} the National Cancer Institute³ and industry⁴ as a new source of novel bioactive natural products. We have therefore collected several cyanophytes as part of a survey of algal chemistry from the north coast of Puerto Rico and we report here the structure of a novel cytotoxin from one of these, Hormothamnion enteromorphoides. A CHCl₃/MeOH extract of fresh H. enteromorphoides demonstrated slight gram positive antimicrobial activity and showed the presence of a distinctive orange-charring (H₂SO₄) yellow pigment by thin layer chromatography.⁵ The readily isolable pigment band contained a subtle mixture of several compounds, the major of which, 1, was isolated by careful reverse phased HPLC and its novel styrylchromone structure solved by x-ray crystallographic means. Trivially named hormothamnione, metabolite 1 is a potent cytotoxin to several human cancerous cell lines and appears to operate via inhibition of RNA synthesis (Table 1).

Senescent yellow-colored H. enteromorphoides was found in abundance on an exposed shallow reef (1-3m) at Playa de Vega Baja, Puerto Rico several times during the months of July and August in 1984 and 1985. The frozen or alcohol preserved seaweed was extracted in standard fashion (CHCl₃/MeOH 2:1) and the yellow pigment band containing 1 was isolated by vacuum chromatography over TLC grade silica gel. Preparative thick layer chromatography of this yellow band simplified the mixture to two components of 75:25 ratio. Small quantities of 1 were obtained from this mixture employing painstaking reverse phased HPLC conditions.⁶



The major component (1) was more conveniently isolated from this mixture by first derivatizing to 2 (Ac₂O, pyr, 16 hrs, RT) and then employing normal phased HPLC conditions.⁷ The natural product 1 was easily reformed (as determined by TLC, MS, NMR) via mild base treatment (K₂CO₃ in MeOH) of 2.

The natural product (1) was a yellow solid (m.p. 270°C dec) which analyzed for C₂₁H₂₀O₈ by high resolution electron impact mass spectrometry (M⁺ M/Z 400.1163, 46%) which upon derivatization to 2 increased in mass to C₂₇H₂₆O₁₁ (M⁺ = base peak at m/z 526 by low resolution CIMS in the negative ion mode) and thus, 2 was the triacetate derivative of 1. The natural product had typical chromone⁸ UV absorptions at 295 and 353 nm (MeOH; ε=6,200; 8,900 respectively) and IR absorptions (KBr) broadly centered at 3400 cm⁻¹ (phenol) and at 1640 cm⁻¹ (pyrone carbonyl). The ¹H NMR (80 MHz, CDCl₃) of 2 confirmed its triacetate nature [δ 2.34 (6H,s) and 2.50 (3H,s)].

Of the remaining seventeen hydrogens in the ¹H NMR spectrum of 2, twelve were present as methyl singlets, three at shifts consistent with aromatic methoxyls [δ 3.89 (3H,s), 4.06 (3H,s) and 4.11 (3H,s)] and one consistent with a deshielded olefinic methyl group [δ 2.17 (3H,s)]. The other 5 protons were separated into two deshielded spin systems, one consistent with a trans disubstituted olefin [δ 7.08 (1H,d,J=16) and δ 7.58 (1H,d,J=16)] and the other interpreted as forming the proton component of a symmetrical 1,3,5 trisubstituted aromatic ring [δ 6.94 (1H,t,J=1.9) and 7.23 (2H,d,J=1.9)]. The nature of the ring system and placement of these proton, methyl, methoxy and acetoxy substituents was not readily determinable, and more definitive NMR experiments were precluded due to limited resources of compound. Hence, an x-ray experiment was performed to unambiguously assign the structure of crystalline 2 (mp 198–201°C).

Preliminary x-ray photographs of the triacetate 2 displayed monoclinic symmetry and accurate lattice constants of a=18.472(5), b=5.769(3), c=24.096(6) Å, and β=93.40(2)° were determined from a least squares analysis of fifteen moderate, diffractometer measured 2θ-values. The systematic extinctions and crystal density indicated space group P2₁/a with one molecule of composition C₂₇H₂₆O₁₁ forming the asymmetric unit (Z=4). All diffraction maxima with 2θ ≤ 114° were measured on a computer controlled four-circle diffractometer with graphite monochromated Cu Kα radiation (1.54178 Å) and variable speed, 1° ω-scans. Of the

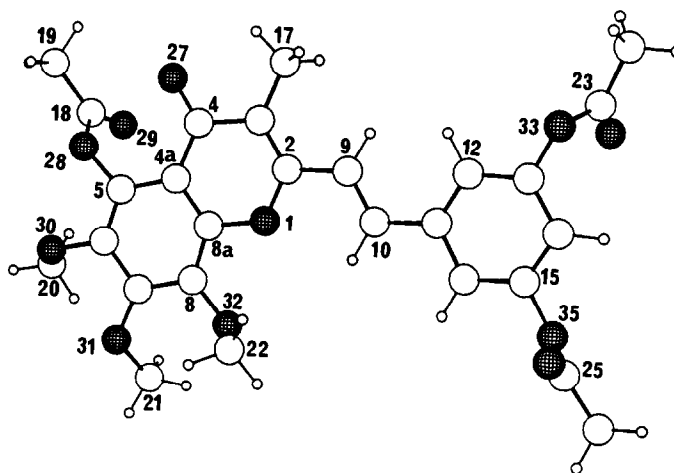


Figure 1. A computer generated perspective drawing of the final x-ray model of hormothamnione triacetate (2). The chromone fragment (atoms O1 to C8a) is planar, and the substituents on the fully substituted aromatic ring are all rotated out of the plane. The relevant torsional angles vary from 50° for C8-C7-O31-C21 to 87° for C7-C8-O32-C22. The acetate substituents on the 1,3,5-substituted ring are also rotated by roughly 77° .

3459 reflections surveyed, only 1211 (35%) were judged observed ($F_o \geq 3\sigma(F_o)$) after correction for Lorentz, polarization and background effects.⁹ The centrosymmetric structure was solved easily. Hydrogen atoms were located on a ΔF -synthesis after partial refinement of the nonhydrogen atoms. Block diagonal least-squares refinements with anisotropic nonhydrogen atoms and isotropic hydrogens have converged to a standard crystallographic residual of 0.081 for the observed reflections. Crystallographic parameters have been deposited with and are available from the Cambridge Crystallographic Data File, University Chemical Laboratory, Lensfield Road, Cambridge, CB2 1EW, England.

The styrylchromone structure of hormothamnione is unprecedented in natural products although the carbon skeleton is known from synthetic studies.¹⁰ It is equally plausible that hormothamnione is biogenetically derived from acetate or shikimate pathways, though the oxidations at C-13 and C-15 are inconsistent with the expected oxidation patterns resulting from either sequence.

As indicated in Table I, the natural product 1 was found to be a potent cytotoxic agent to P388 lymphocytic leukemia and HL-60 human promyelocytic leukemia cell lines. Macromolecule biosynthesis in the presence of 1 was measured via radioactive precursor incorporation studies using HL-60 and KB cell lines (Table I). A major mode of cytotoxic action of hormothamnione appears to be by inhibition of RNA synthesis. Further studies on the site of cytotoxic action of this novel marine natural product are underway.¹¹

Table I. Inhibition of Cell Growth and Macromolecule Biosynthesis in Several Cancerous Human Cell Lines by Hormothamnione (1)

Cell-line	ID ₅₀ (ng/ml) ^a Cell Growth	ID ₅₀ (ng/ml) ^{bc} DNA	ID ₅₀ (ng/ml) ^{bd} RNA	ID ₅₀ (ng/ml) ^{be} Protein
P-388	4.6	---	---	---
HL-60	0.1	300	1.0	> 500
KB	---	530	140	> 500

^aID₅₀ is the drug concentration required to inhibit cell growth to 50% of control levels.

^bThe percentages of macromolecule biosynthesis inhibition were calculated from the specific activities of incorporated precursors (c-e) into 10⁶ cells in drug treated as versus control cells. ID₅₀ values were obtained by plotting the logarithmic drug concentrations against the percentage inhibition of macromolecule biosynthesis.

^cExponentially growing cells pulse labeled with 1 μCi of ³H-thymidine (16.6 Ci/mole) per mL of cell suspension.

^dExponentially growing cells pulse labeled with 2 μCi [5-³H]-uridine (28 Ci/mole) per mL of cell suspension.

^eExponentially growing cells pulse labeled with 1 μCi of ³H-L-leucine (130 Ci/mole) per mL of cell suspension.

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- (4) 1984 Prospectus, Sea Pharm, Inc., Ft. Pierce, Florida.
- (5) Paper disc-agar plate method, using 100 μg and 25 μg of 1 per 6 mm paper disc, tested against Staphylococcus aureus (-), Bacillus subtilus (10 mm zone at 100 μg, 8 mm zone at 25 μg), Escherichia coli (-) and Candida albicans (-). Hormothamnione was inactive in our goldfish toxicity experiments (25 μg/ml H₂O).
- (6) Two Waters 3.9 mm x 25 cm μ-Bondapak C-18 columns in series, 40% H₂O/MeOH as mobile phase, 1.5 ml min, 35 min retention time, 0.5 mg per injection.
- (7) Waters 8 mm x 10 cm μ-Porasil Radial Compression Column, 40% EtOAc/isooctane, 3.0 ml/min, 7 min retention time, 1 mg/injection.
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- (11) Our gratitude goes to Dr. David Ballantine for taxonomic identifications of Hormothamnion enteromorphaeoides. Work at Cornell University was supported by NIH CA24487 and the New York State Sea Grant Program. We acknowledge at Oregon State University the mass spectral facilities in the Department of Agricultural Chemistry run by Dr. Max Deinzer and the NMR facility in the Department of Chemistry. Collecting efforts in Puerto Rico were supported by the Puerto Rico Sea Grant Program while work conducted at Oregon State University was supported by the Oregon Sea Grant Program and the OSU Research Council.

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